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pg. 16 Selected Chemistry Assignment AnswersUnit 6: Thermochemistry, States of Matter and Intermolecular ForcesChapter 10: Causes of Change10.1: Energy Transfer(Practice on pg. 342)1. 97 J 2. 1.3 mol 3. 2.2 102 J 4. 197 K(Section Review on pg. 344)9. 29.2 kJ 10. 14.6 kJ 11. 0.52 mol 12. 358 K13. 295 K 14. 0.234 J (K g) 15. 0.864 J (K g) 18. 100 J (K mol)19. 0.6 J (K g)10.2: Using Enthalpy(Practice...

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pg. 20 Selected Honour Chemistry Assignment AnswersUnit 5: Chemical Kinetics and EquilibriumChapter 14: Chemical Kinetics (pg. 489 to 493)27. Activation energy is the amount of energy needed to overcome for the products to form. All reactions have some kind of activation energy. The higher is the activation energy, the slower the reaction rate.30. Methane and oxygen gas mixture does not react spon...

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[UNIT 4 ASSIGNMENTS ANSWERS](#)

pg. 16 Selected Honour Chemistry Assignment AnswersUnit 4: Thermochemistry and Nuclear ChemistryChapter 6: Energy Relationships in Chemical Reactions (pg. 198 to 202)1. System: - any specific part of the universe that is of interest to us. Surrounding: - the rest of the universe outside the system. Open System: - a system that can exchange mass and energy (usually in the form of heat) with its sur...

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