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a.  $\text{Cu (s)} + \text{O}_2 \text{ (g)} \rightarrow \text{CuO (s)}$   
b.  $\text{H}_2\text{O (l)} \rightarrow \text{H}_2 \text{ (g)} + \text{O}_2 \text{ (g)}$   
c.  $\text{Fe (s)} + \text{H}_2\text{O (g)} \rightarrow \text{H}_2 \text{ (g)} + \text{Fe}_3\text{O}_4 \text{ (s)}$   
d.  $\text{AsCl}_3 \text{ (aq)} + \text{H}_2\text{S (aq)} \rightarrow \text{As}_2\text{S}_3 \text{ (s)} + \text{HCl (aq)}$   
e.  $\text{CuSO}_4 \cdot 5 \text{H}_2\text{O (s)} \rightarrow \text{CuSO}_4 \text{ (s)} + \text{H}_2\text{O (l)}$

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